

EnzyChrom™ Glucose Oxidase Assay Kit (EGOX-100)

Quantitative Colorimetric/Fluorimetric Glucose Oxidase Activity Determination

DESCRIPTION

Glucose oxidase catalyzes the oxidation of glucose from D-glucose to D-glucono-δ-lactone. Physiologically, it aids in the breakdown of glucose into smaller metabolites. It is widely used in electrochemical glucose sensors designed for diabetes patients. Simple, direct and high-throughput assays for measuring glucose oxidase activity find wide applications in research and drug discovery. BioAssay Systems' glucose oxidase assay kit uses a single Working Reagent that combines the glucose oxidase reaction and color reaction in one step. The change in color intensity of the reaction product at 570 nm or fluorescence intensity at $\lambda_{ex/em} = 530/585$ nm is directly proportional to glucose oxidase activity in the sample.

KEY FEATURES

Sensitive and accurate. Use as little as 20 μ L samples. Linear detection range in 96-well plate for 20 minute incubation at 25°C: 0.02 to 10 U/L glucose oxidase for colorimetric assays and 0.002 to 1.5 U/L for fluorimetric assays.

Simple and high-throughput. The procedure involves addition of a single working reagent and incubation for 20 min at room temperature.

APPLICATIONS

Direct Assays: glucose oxidase activity in cell lysate, culture medium and other biological samples.

Drug Discovery/Pharmacology: effects of drugs on glucose metabolism.

KIT CONTENTS (100 tests in 96-well plates)

Assay Buffer: 10 mL **HRP Enzyme:** 120 μ L
Glucose: 1.5 mL 2 M Glucose **Dye Reagent:** 120 μ L
Standard: 100 μ L 3% H₂O₂

Storage conditions. The kit is shipped on ice. Store all components at -20°C. Shelf life: 6 months after receipt.

Precautions: reagents are for research use only. Normal precautions for laboratory reagents should be exercised while using the reagents. Please refer to Material Safety Data Sheet for detailed information.

COLORIMETRIC PROCEDURE

Samples can be analyzed immediately after collection, or stored in aliquots at -20 °C. Avoid repeated freeze-thaw cycles. If particulates are present, centrifuge sample and use clear supernatant for assay.

1. Equilibrate all components to room temperature. During experiment, keep thawed Enzyme in a refrigerator or on ice.
2. **H₂O₂ Standard Curve.** Mix 5 μ L 3% H₂O₂ and 914 μ L dH₂O (final 4.8 mM) then mix 20 μ L of the 4.8 mM H₂O₂ with 460 μ L dH₂O to yield 200 μ M H₂O₂. Prepare standards as shown in the Table below.

No	400 μ M H ₂ O ₂ + H ₂ O	Vol (μ L)	H ₂ O ₂ (μ M)
1	100 μ L + 0 μ L	100	200
2	60 μ L + 40 μ L	100	120
3	30 μ L + 70 μ L	100	60
4	0 μ L + 100 μ L	100	0

Transfer 20 μ L standards and samples into separate wells.

3. **Working Reagent.** Prepare bulk working reagent by mixing 75 μ L Assay Buffer, 10 μ L 2 M Glucose, 1 μ L HRP Enzyme (*vortex briefly before pipetting*), and 1 μ L Dye Reagent per reaction well in a clean tube. Transfer 80 μ L Working Reagent into each reaction well. Tap plate to mix.
4. Read optical density immediately (OD₀) at 570 nm (550-585 nm). Incubate 20 min at room temperature, and then read optical density again (OD₂₀).

FLUORIMETRIC PROCEDURE

For fluorimetric assays, the linear detection range is 0.002 to 1.5 U/L glucose oxidase. Dilute the standards from *Colorimetric Procedure* 10 \times with dH₂O to obtain standards at 20, 12, 6 and 0 μ M H₂O₂.

Transfer 20 μ L standards and 20 μ L samples into separate wells of a *black* 96-well plate.

Add 80 μ L Working Reagent (see *Colorimetric Procedure*), tap plate to mix.

Read fluorescence immediately (F₀) at $\lambda_{ex/em} = 530/585$ nm, incubate 20 min at room temperature, and then read fluorescence again (F₂₀).

CALCULATION

Subtract blank OD₂₀ or F₂₀ (water, #4) from all standard OD₂₀ or F₂₀ values and plot the Δ OD or Δ F against standard concentrations. Determine the slope using linear regression. Calculate the Δ OD_{Sample} or Δ F_{Sample} of all samples by subtracting OD₀ or F₀ from OD₂₀ or F₂₀ for each sample. Do the same for the blank (water, standard #4) to get Δ OD_{Blank} or Δ F_{Blank}. Calculate the activity using the equation below:

$$GO \text{ Activity} = \frac{\Delta R_{\text{SAMPLE}} - \Delta R_{\text{BLANK}}}{\text{Slope } (\mu\text{M}^{-1}) \cdot t} \times n \text{ (U/L)}$$

Where ΔR_{Sample} and ΔR_{Blank} are the change in optical density or fluorescent values of the sample and blank, respectively. *Slope* is the slope of the H₂O₂ standard curve, *t* is the incubation time (20 minutes), and *n* is the dilution factor.

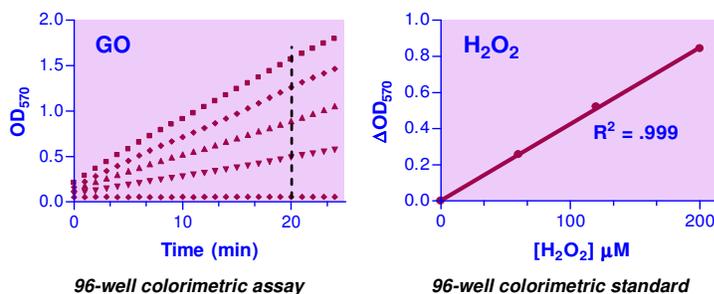
Notes: If the calculated sample glucose concentration is higher than 10 U/L in colorimetric assay or 1.5 U/L in fluorimetric assay, dilute sample in water and repeat the assay. Multiply result by the dilution factor (*n*). For samples with low Glucose Oxidase activity, the incubation time can be increased.

Unit definition: 1 U/L of Glucose Oxidase catalyzes 1 μ mole of H₂O₂ per minute at pH 7.0 and room temperature.

MATERIALS REQUIRED, BUT NOT PROVIDED

Pipetting devices, centrifuge tubes, clear flat-bottom 96-well plates, black 96-well plates and plate reader.

Glucose Oxidase Graphs



LITERATURE

1. Raba J, Mottola HA. (1995). Glucose Oxidase as an Analytical Reagent. *Critical Reviews in Analytical Chemistry* 25(1):1-42.
2. Harris JM, Reyes C, Lopez GP. (2013). Common Causes of Glucose Oxidase Instability in in vivo Biosensing: a Brief Review. *J Diabetes Sci Technol* 7(4):1030-8.
3. Ferri S, Kojima K, Sode K. (2011). Review of glucose oxidases and glucose dehydrogenases: a bird's eye view of glucose sensing enzymes. *J Diabetes Sci Technol* 5(5):1068-76.

